

# Variables

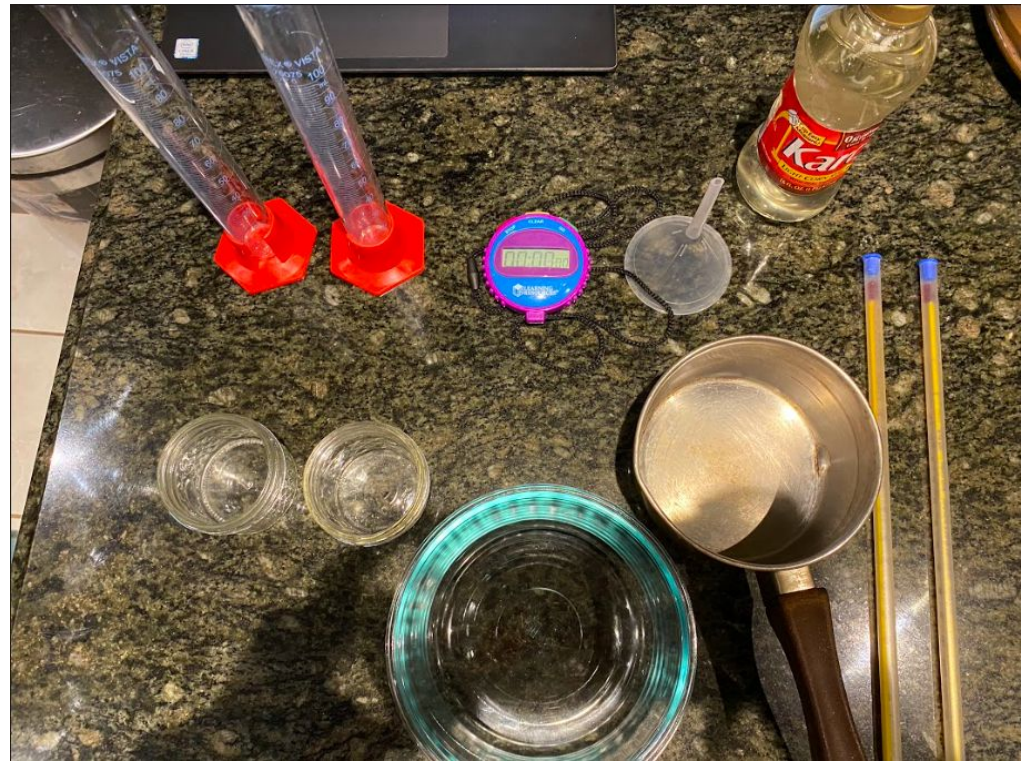
- ◉ **Independent variable:** My independent variable is the temperature of the corn syrup. I changed the independent variable by increasing the temperature to 140 °F. I also decreased the temperature to 45 °F.
- ◉ **Dependent variable:** My dependent variable is the viscosity of the corn syrup after the temperature changes.
- ◉ **How I Collected My Data:** I measured the dependent variable by timing the amount of time it takes for the corn syrup to pass through a funnel after each temperature change. I repeated my measurements 10 times to be more reliable.

# Controlled Variables

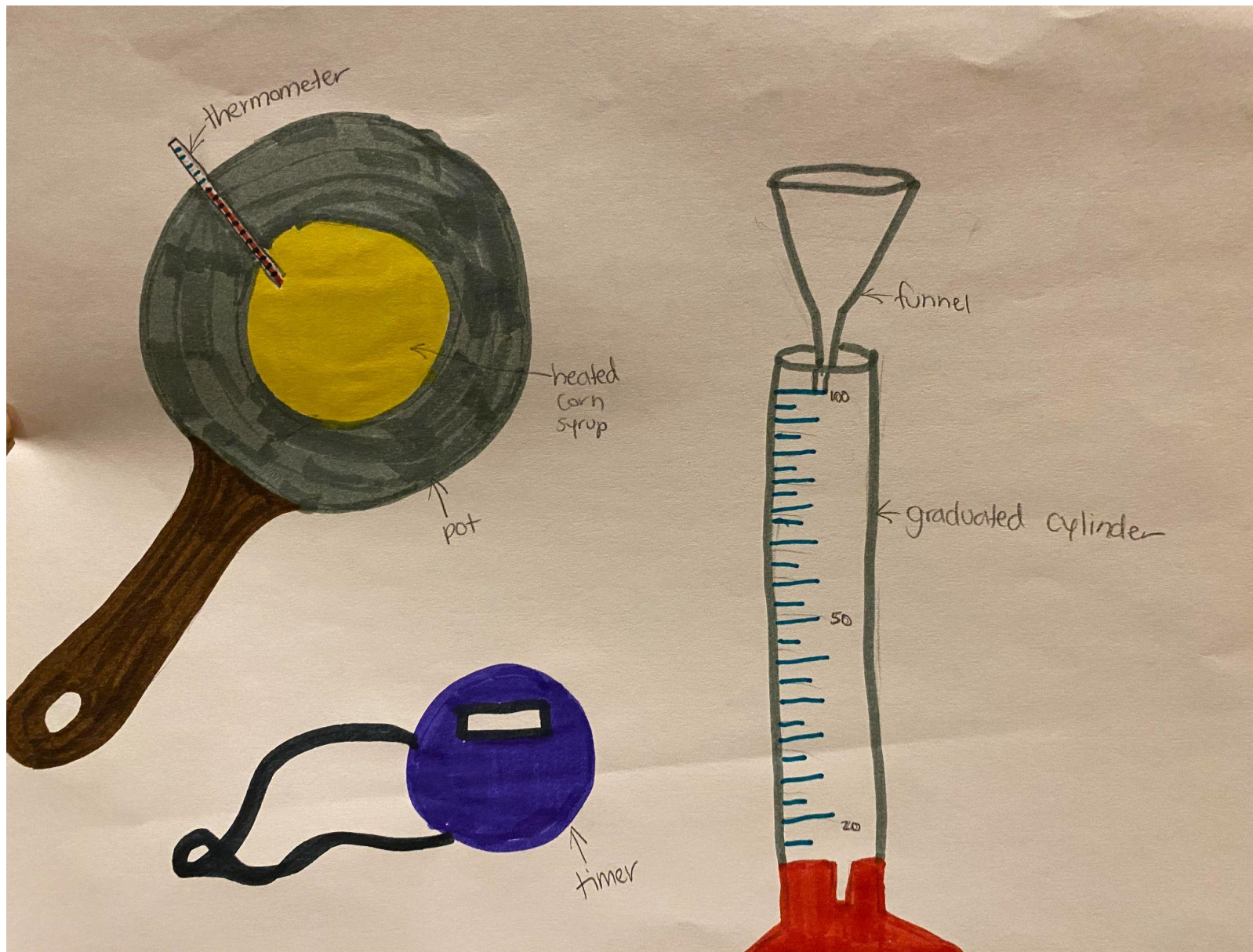
| Controlled Variables  | Why It Needs to be Controlled  |
|---|--|
| The amount of liquid being tested   | It will take longer for more liquid to pass through the funnel, when compared to less liquid |
| The funnel being used   | More liquid will be able to pass through the funnel, if the size of the funnel changes       |
| The temperature the corn syrup is being heated to                             | The viscosity of the liquid will change at different temperatures                            |
| The temperature the corn syrup is being cooled to                             | The viscosity of the liquid will change at different temperatures                            |
| The 'Room Temperature' temperature  | The viscosity of the liquid will change at different temperatures                            |
| The amount of time the corn syrup is left out after heating and before timing | Different amounts of time between heating and measuring will affect viscosity                |
| The amount of time the corn syrup is left out after cooling and before timing | Different amounts of time between cooling and measuring will affect viscosity                |
| The thermometer being used  | The thermometer might be read differently, depending on thermometer                          |

# Materials

- *One bottle of Karo brand light corn syrup*
- *One pot*
- *Two thermometers*
- *A freezer*
- *Small containers (I used Mason Jars)*
- *Four funnels*
- *Four Graduated cylinders*
- *A big bowl*
- *Four Timers*
- *Heat Source*







**Labeled Diagram**



## Photos!



Room Temperature Corn Syrup  
Going Through the Funnel



Cold Corn Syrup Going Through  
the Funnel



Room Temperature Corn Syrup  
Going Through the Funnel

# Data Table

**Table**

|       | Trial 1 | Trial 2 | Trial 3 | Trial 4 | Trial 5 | Trial 6 | Trial 7 | Trial 8 | Trial 9 | Trial 10 |
|-------|---------|---------|---------|---------|---------|---------|---------|---------|---------|----------|
| Temp. | 140     | 140     | 139     | 139     | 140     | 139     | 140     | 140     | 139     | 140      |
| Time  | 1:49    | 00:35   | 1:22    | 1:57    | 1:35    | 1:25    | 00:55   | 1:38    | 1:29    | 1:15     |
| Temp. | 45      | 45      | 45      | 45      | 45      | 45      | 45      | 45      | 45      | 45       |
| Time  | 39:00   | 48:05   | 42:25   | 48:42   | 45:52   | 47:24   | 40:54   | 44:43   | 46:08   | 43:13    |
| Temp. | 70      | 70      | 70      | 70      | 70      | 70      | 70      | 70      | 70      | 70       |
| Time  | 25:34   | 27:54   | 22:38   | 20:35   | 21:45   | 11:24   | 20:00   | 15:24   | 15:35   | 20:00    |

Key:

Heated

Cooled

Room Temperature

Observations:

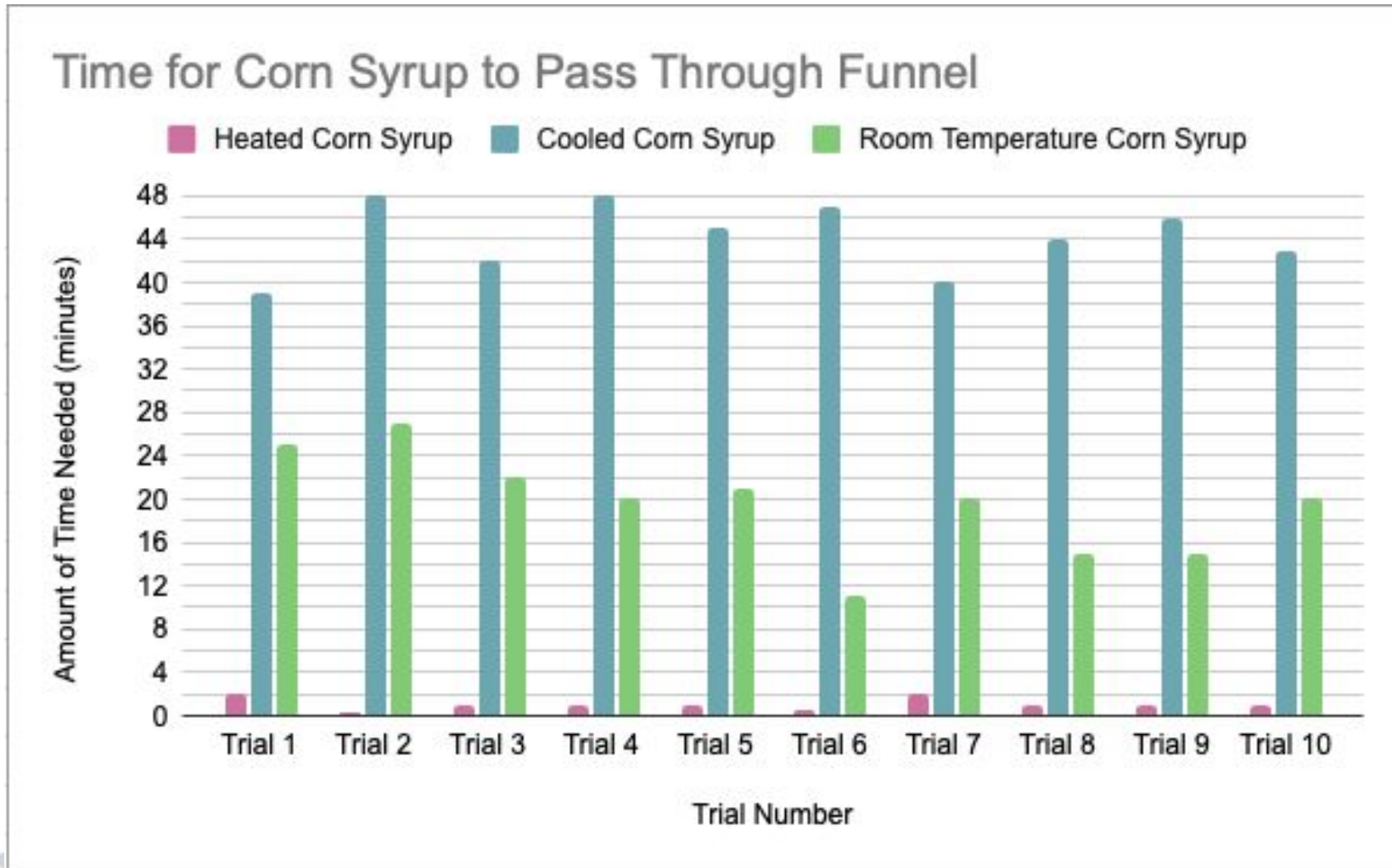
In Trial 1, the corn syrup boiled at 65 degrees and developed a film on top.

It heats up very quickly.

After 15 minutes, the cold corn syrup increased in temperature by 15 degrees.

Trial 6 melted the spatula I was using to stir, so I redid it.

# Graph



# Analysis

- © My data shows that when corn syrup is cooled, its viscosity increases and when corn syrup is heated, its viscosity decreases. My data suggests that viscosity is directly related to temperature of certain liquids. This might be because molecules move slower when the liquid is colder.



# Summary

- I conclude that this experiment has helped me solve my original problem. This is because I was able to get the data I wanted to find about viscosity and temperature, from this experiment.
- I predicted that when the corn syrup is cooled, the viscosity would increase. My data supports my prediction. I think this is because I had previous knowledge about molecules and how they act in different temperatures. The viscosity was higher in colder temperatures and lower in higher temperatures because of the movement of the molecules.

# Future Work

- ◎ The method I followed did allow me to answer the research question. I think this is because I was able to collect the data I needed to answer the research question. Some strengths in the method were the data it allowed me to collect and the simplicity of the experiment. One weakness in the method was the amount of time it took. Something I found difficult in carrying out the method was making sure I checked and stopped the timer at the right time, because otherwise my experiment wouldn't have been valid. If I wanted to test the same problem again, I would use the same method. This is because I was able to get the data I needed and it was simple and easy to follow.
- ◎ I could improve the method by decreasing the amount of trials or measuring the viscosity in a different way. I would make these improvements because it would save time and it would still be simple. This investigation has made me think of a new question, which is how would adding water affect the viscosity of a liquid. I could test this the same way but adding water to the liquid in the beginning. I would like to find out more about what affects viscosity because I know that water reacts differently when cooled.